Balancing the equation see-saw

A class has handed in their homework. The students had to balance equations. Now, as their teacher, you have to mark the homework. Unfortunately, you find that you have misplaced your answer sheet. It seems like you have to balance the equations yourself! Remember to show all your working.

Here is a copy of the worksheet.

Equations to balance

(i) iron + sulfur \rightarrow iron(II) sulfide

$$Fe + S \rightarrow FeS$$

(ii) magnesium oxide + sulfuric acid → magnesium sulfate+water

$$MgO + H_2SO_4 \rightarrow MgSO_4 + H_2O$$

(iii) hydrogen + oxygen → water

$$H_2 + O_2 \rightarrow H_2O$$

(iv) calcium carbonate + sulfuric acid → calcium sulfate + carbon dioxide + water

$$CaCO_3 + H_2SO_4 \rightarrow CaSO_4 + CO_2 + H_2O_3$$

(v) sulfuric acid + sodium hydroxide → sodium sulfate + water

$$H_2SO_4 + NaOH \rightarrow Na_2SO_4 + H_2O$$

(vi) ammonia + nitric acid → ammonium nitrate

(vii) potassium iodide + chlorine → potassium chloride + iodine

$$KI + Cl_2 \rightarrow KCI + l_2$$

(viii) zinc + copper(II) sulfate → zinc sulfate + copper

$$Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$$

(ix) aluminium + iron(III) oxide → aluminium oxide + iron

$$AI + Fe2O3 \rightarrow Al2O3 + Fe$$